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Worksheet 5 Double Replacement Reactions

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Worksheet 5 Double Replacement Reactions

Worksheet #5: Double-Replacement Reactions In these reactions, all you do is look at the names of the reactants, and "switch partners". Just be sure that the new pairs come out with the positive ion named first, and paired with a negative ion. 1. aluminum iodide + mercury(II) chloride \rightarrow 2.

Worksheet #5: Double-Replacement Reactions In these ...

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Worksheet #5: Double-Replacement Reactions In these reactions, all you do is look at the names of the reactants, and "switch partners". Just be sure that the new pairs come out with the positive ion named first, and paired with a negative ion. 1.

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One example of a double displacement reaction is the reaction of lead (II) nitrate with ... single replacement 4) double replacement 5) acid-base (double replacement) 6) synthesis 7) decomposition . Practice Balancing Chemical Equations Worksheet . Balance the following chemical equations using coefficients . 1 . 1. $\text{Al(OH)}_3 (\text{s}) + 3.$

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Unit 5 Chemical Equations and Reactions

Practice Problem Answers. Back to Double Replacement. Write correct formulas for the products in these double replacement reactions.

- 1) $\text{Ca}(\text{OH})_2 + \text{H}_3\text{PO}_4 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$
- 2) $\text{K}_2\text{CO}_3 + \text{BaCl}_2 \rightarrow \text{KCl} + \text{BaCO}_3$
- 3) $\text{Cd}_3(\text{PO}_4)_2 + (\text{NH}_4)_2\text{S} \rightarrow \text{CdS} + (\text{NH}_4)_3\text{PO}_4$
- 4) $\text{Co}(\text{OH})_3 + \text{HNO}_3 \rightarrow \text{Co}(\text{NO}_3)_3 + \text{H}_2\text{O}$
- 5) $\text{AgNO}_3 + \text{KCl} \rightarrow \text{AgCl} + \text{KNO}_3$
- 6) $\text{Na}_2\text{CO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \dots$

ChemTeam: Double Replacement Problem Answers

Showing top 8 worksheets in the category - Displacement Reaction. Some of the worksheets displayed are Chemistry single replacement reaction work, Work 4 single replacement reactions step 1, 46 5 displacement reactions, Chemistry replacement reaction work, Chemistry single replacement reaction work, Work 6 displacement reactions and acidbase

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Double Displacement Reactions - Lesson Worksheets

Replacement Worksheet-Chem 121 12) $\text{Ag}_2\text{S} + \text{CuCl}_2 \rightarrow$ 13) $\text{Pb}(\text{NO}_3)_2 + \text{HI} \rightarrow$ 14) $\text{Ba}(\text{ClO}_3)_2 + \text{H}_2\text{SO}_4 \rightarrow$ 15) $\text{CuS} + \text{KCl} \rightarrow$ SINGLE REPLACEMENT REACTION DESCRIPTION In these reactions, a free element reacts with a compound to form another compound and release one of the elements of the original compound in the elemental

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state. There are ...

CHEMISTRY REPLACEMENT REACTION WORKSHEET

Single Replacement Reaction C

Synthesis Reaction A. Double

Replacement Reaction D Decomposition

Reaction B. 7. Can two elements be used as reactants for a synthesis reactions? If yes, give at least one example from

Model 1 to support your answer. Yes,

$4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$ 8.

Types of Chemical Reactions Key - Google Docs

05 - Worksheet_Double Displacement Rxn.pdf ... Loading...

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5 Main Patterns of Reactions * Synthesis

* Decomposition * Single Displacement *

Double Displacement * Combustion

Synthesis - To ... Also known as double

replacement; Usually occurs between

two ionic compounds, which then in turn

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create two more ionic compounds.

Types of Reactions - Synthesis, Decomposition, Single and ...

Artwork taken from an unknown source (a worksheet acquired second-hand many years ago). Types of Reactions Chemical reactions can be categorized into a variety of types. Five common types that we will study are synthesis, decomposition, combustion, single replacement and double replacement. Be familiar

Types of Reactions - gbschemphys

Classifying Chemical Reactions

Worksheet Name: _____ Period: _____

Classify each reaction as acid/base, redox, synthesis, decomposition, single replacement, double replacement or combustion. They may be more than one. The equations are not balanced. 1.

Classifying Chemical Reactions Worksheet

NaOH. Type of reaction: double

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replacement. 2) $2. \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$. Type of reaction: synthesis.
3) 4. $\text{C}_5\text{H}_9\text{O} + 27$. Six Types of Chemical Reaction Worksheet Compilation of the 5 Types Chemical Reactions. Word equations included for all reactions.
UPDATE: Synthesis Rxn- Word Equation:
Iron(II) + Sulfur yields Iron(II) Sulfide
Page 8/11.

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